This article was downloaded by:

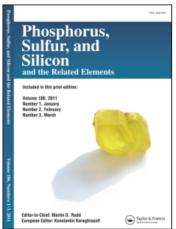
On: 29 January 2011

Access details: Access Details: Free Access

Publisher Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3W, LW

41 Mortimer Street, London W1T 3JH, UK



### Phosphorus, Sulfur, and Silicon and the Related Elements

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713618290

# Cyclic and Bicyclic (Ph<sub>2</sub>PCPX) Oligomers

Alfred Schmidpetera; Hans-Peter Schrödela; Georg Jochema

<sup>a</sup> Institut für Anorganische Chemie der Universität München Meiserstrasse 1, München, Germany

To cite this Article Schmidpeter, Alfred , Schrödel, Hans-Peter and Jochem, Georg(1994) 'Cyclic and Bicyclic ( $Ph_3PCPX$ ) Oligomers', Phosphorus, Sulfur, and Silicon and the Related Elements, 93: 1, 321 - 324

To link to this Article: DOI: 10.1080/10426509408021845 URL: http://dx.doi.org/10.1080/10426509408021845

### PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

# Cyclic and Bicyclic (Ph<sub>3</sub>PCPX) Oligomers

Alfred Schmidpeter, Hans-Peter Schrödel and Georg Jochem

Institut für Anorganische Chemie der Universität München Meiserstrasse 1, D-80333 München, Germany

The title compounds  $(Ph_3PCPX)_n$ , X = Cl, Br and n = 2, 3, 4 are obtained from the condensation of a disilylylide and phosphorus trichloride or tribromide  $^{1-3}$ .

They are analogues in composition to the cyclophosphazanes (RNPX)<sub>n</sub> with the imino groups RN exchanged for the isoelectronic triphenylphosphonium ylide groups Ph<sub>3</sub>PC. The ylide moieties exert, however, a definitely stronger electronic effect and are thus expected to shift the systems more towards dissociation of the PX bonds. This must necessarily have its structural consequences.

According to their A<sub>2</sub>B<sub>2</sub> type <sup>31</sup>P-NMR spectra and also to an X-ray investigation <sup>4</sup> the dimers (Ph<sub>3</sub>PCPX)<sub>2</sub> have the expected 1,3-diphosphetane structure. The PCl bonds are extremely long (220 and 225 pm) because of the strong negative hyperconjugation in the present conformation. Chloride abstraction, nucleophilic substitution, 1,3-dipolar cycloaddition and reductive ring enlargement reactions give rise to new cationic four- and five-membered heterocycles of two-coordinate phosphorus.

$$Ph_{3}P \xrightarrow{P} PPh_{3} \xrightarrow{Me_{3}SiN_{3}} Ph_{3}P \xrightarrow{P} PPh_{3}$$

$$Ph_{3}P \xrightarrow{P} PPh_{3}$$

The <sup>31</sup>P-NMR spectra of the trimers (Ph<sub>3</sub>PCPX)<sub>3</sub> also at first seem to indicate a symmetric structure. They turn out, however, to result from a rapid exchange of ionic and covalent PX bonds. At lower temperatures or when the ionic halide is replaced by tosylate the spectra show all six phosphorus atoms to be nonequivalent. This suggests a structure of the cation as for the isoelectronic S<sub>4</sub>N<sub>2</sub> molecule <sup>5</sup>, with the two-coordinate phosphorus atom bonded like in the acyclic prototype <sup>1</sup> and with the remaining two PX bonds in equatorial and axial orientation.

If the preparation of the bromide is performed under a little different conditions instead of the trimer (Ph<sub>3</sub>PCPBr)<sub>3</sub> a product with one ylide moiety less (Ph<sub>3</sub>PC)<sub>2</sub>(PBr)<sub>3</sub> is obtained.

SiMe<sub>3</sub>

$$Ph_{3}P$$

By <sup>31</sup>P-NMR spectra and X-ray investigation <sup>4</sup> it is identified as a 1,2,4-triphospholane derivative. Of the three PBr bonds that at P-4 is longest (247 pm) and is dissociated in solution. Reduction leads to the 1,2,4-triphosphole cation mentioned above.

In the chloride series we could also obtain the tetramer (Ph<sub>3</sub>PCPCl)<sub>4</sub> which, however, has not the expected eight-membered ring structure. It is ionic both in the crystal and in solution, the cation in contrast to all the other cations mentioned before does not contain a two-coordinate phosphorus. As seen from the <sup>31</sup>P-NMR spectra and from an X-ray investigation <sup>4</sup> it has a threefold symmetry. The remaining PCl bonds (235 pm) are still longer than in the dirner.

If one chlorine substitutent in the cation is replaced for an amino group, a second dissociation takes place, the cation rearranges as to provide the new two-coordinate phosphorus the same stabilisation as in the trimer.

Summary: The isoelectronic RN/Ph<sub>3</sub>PC exchange leads from the cyclophosphazanes to a novel series of phosphorus heterocycles (Ph<sub>3</sub>PCPX)<sub>n</sub> which are analogous in composition but mostly different in structure.

$$(RNPX)_{n} \qquad Ph_{3}P \Rightarrow PPh_{3}$$

$$Ph_{3}P \Rightarrow PPh_{3}$$

The tendency for dissociation is higher and increases with n. In the cation of the trimer three ylide moieties share the positive charge in a dynamic mode, in the cation of the tetramer they form bicyclic structures in which the additional bond compensates the loss of electrons and eventually allows a second dissociation.

Acknowledgement: We thank the Deutsche Forschungsgemeinschaft and the Fonds der Chemischen Industrie for support.

#### References

- 1. A. Schmidpeter and G. Jochem, Tetrahedron Lett. 33, 471 (1992).
- A. Schmidpeter, G. Jochem and M. Thiele, *Phosphorus, Sulfur, Silicon*, 76, 13 (1993).
- 3. A. Schmidpeter, H. Nöth, G. Jochem, H.-P. Schrödel and K. Karaghiosoff, *Chem. Ber.* in press.
- 4. X-Ray investigations by H. Nöth.
- R.N.H. Small, A.J. Banister and Z.V. Hauptman, J. Chem. Soc. Dalton Trans. 1981, 2188; T. Chivers, P.W. Codding, W.G. Laidlow, S.W. Liblong, R.T. Oakley and M. Trsic, J. Am. Chem. Soc., 105, 1186 (1983).